# Ipvelutine, 7 $\beta$-Acetoxy-2 $\alpha$-(tigloyloxy)tropane, an Unusual Tropane Alkaloid from Ipomoea velutina R. Br. (Convolvulaceae) 

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#### Abstract

Convolvulaceae provide a rich source of tropane alkaloids, however, 2-substituted tropanes have been described for only few species of this taxon. In this note, 2,7 -diesters such as ipvelutine [ $7 \beta$-acetoxy-2 $\alpha$-(tigloyloxy)tropane] isolated from the vegetative parts of the Australian Ipomoea velutina R. BR. are described as a new group of tropane diesters.


## Keywords

Ipomoea velutina • Convolvulaceae • Ipvelutine • 7 $\beta$-Acetoxy-2 $\alpha$-tigloyloxytropane • 2,7-Disubstituted Tropanes • Structure Elucidation

## Introduction

During our continuous studies on secondary metabolites of the Convolvulaceae, this plant family has been shown to produce a plethora of tropane alkaloids, especially 3-tropanols and their esters (e.g. [1, 2]), as well as some 3,6-disubstituted tropanes [3] or the polyhydroxylated calystegines [4]. This underlines the chemotaxonomic relationship with their sister family Solanaceae where the biosynthetic pathway of tropane alkaloids is well investigated. The main route leads to two stereoisomeric 3-hydroxytropanes, namely
$3 \alpha$-tropanol (basic component of the well-known atropine and other esters), and $3 \beta$-tropanol which is also precursor of the calystegines. 2-Substituted tropane alkaloids could only be found as a by-product in the Solanaceae [5]. Accordingly, amongst the tropane alkaloids of the Convolvulaceae 2 -substituted ones are extremely rare, too, and could only be detected in some Calystegia, Erycibe, and Ipomoea species [6].

## Results and Discussion

In the alkaloidal screening of Convolvulaceae via GC-MS analysis the basic extracts of the Australian Ipomoea velutina R. Br. revealed the presence of several unknown substances. In the basic extract of the vegetative parts seven unknown nitrogen-containing compounds were detected: one main alkaloid and six minor ones ( $0.7-18.7 \%$ of the main alkaloid by integration of the corresponding GC-MS peaks). The molecular formula of the main compound (1) is consistent with $\mathrm{C}_{15} \mathrm{H}_{23} \mathrm{NO}_{4}(\mathrm{~m} / \mathrm{z} 281)$.

The ${ }^{1} \mathrm{H}-\mathrm{NMR}$ (Table 1) in combination with HSQC and HMBC experiments showed two acylic residues: a $\mathrm{C}_{5}$-acid containing a double bond, namely tiglic acid, as well as acetic acid. Both were confirmed by fragmentation ions in the EIMS as products of $\alpha$-cleavage neighbouring the ester carbonyls: $\mathrm{m} / \mathrm{z} 83\left(\mathrm{C}_{4} \mathrm{H}_{7}-\mathrm{CO}^{+}\right.$; HRMS: $\left[\mathrm{C}_{5} \mathrm{H}_{7} \mathrm{O}\right]^{+}$as 83.04959, calcd. 83.04969) and $m / z 43\left(\mathrm{CH}_{3}-\mathrm{CO}^{+}\right)$.

Tab. 1. $\quad{ }^{1} \mathrm{H}$ - and ${ }^{13} \mathrm{C}$-NMR data of ipvelutine (in MeOD)

| atom | $\begin{gathered} 1 \mathrm{H}-\mathrm{NMR} \\ \text { (in MeOD) } \end{gathered}$ |  |  | $\begin{aligned} & { }^{13} \mathrm{C}-\mathrm{NMR}{ }^{*} \\ & \text { (in MeOD) } \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: |
| 1 | 3.55 | brd | 3.2 Hz | 72.9 |
| 2a | 5.02 | ddd | 2.2 Hz; 5.9 Hz; 11.3 Hz | 68.9 |
| 3 e | 1.98 | $m$ |  | 22.8 |
| 3 a | 1.49 | $d t d$ | 6.4 Hz; 12.1 Hz; 12.8 Hz | 22.8 |
| 4a | 1.89 | $m$ |  | 27.5 |
| 4 e | 1.68 | ddd | 2.3 Hz; 6.7 Hz; 13.7 Hz | 27.5 |
| 5 | 3.82 | br $t$ | 5.2 Hz | 64.7 |
| 6 n | 2.36 | dd | $8.0 \mathrm{~Hz} ; 14.6 \mathrm{~Hz}$ | 37.8 |
| 6 x | 2.27 | ddd | 3.5 Hz ; 6.3 Hz ; 14.7 Hz | 37.8 |
| 7n | 4.61 | dd | $3.4 \mathrm{~Hz} ; 7.9 \mathrm{~Hz}$ | 70.8 |
| $\mathrm{N}-\mathrm{CH}_{3}$ | 2.91 | s |  | 40.9 |
| 1' |  |  |  | 167.8 |
| 2' |  |  |  | 129.0 |
| 3' | 6.96 | $d q$ | 1.2 Hz; 6.9 Hz | 139.3 |
| $\mathrm{CH}_{3}-4{ }^{\prime}$ | 1.83 | d | 7.1 Hz | 11.9 |
| $\mathrm{CH}_{3}-5{ }^{\prime}$ | 1.84 | $d$ | 0.9 Hz | 14.1 |
| $1{ }^{\prime \prime}$ |  |  |  | 176.7 |
| $\mathrm{CH}_{3}-2$ " | 1.93 | S |  | 21.9 |

The HSQC spectrum revealed a characteristically downfield shifted $\mathrm{N}-\mathrm{CH}_{3}\left(\delta_{\mathrm{C}} 40.9, \delta_{\mathrm{H}}\right.$ 2.91) as well as three methylene signals ( $\delta_{C} 37.8,27.5$, and 22.8) and four methine groups
( $\delta_{C} 72.9,70.8,68.9$, and 64.7 ). From the ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}-\mathrm{COSY}$, the complete coupling sequence could be deduced. As a result, 1 (Fig. 1) could be identified as a 2,7-disubstituted tropane.

The substitution pattern of the tropane diester was derived from the mass spectrometric data on the basis of the specific mass fragmentation in bridge-substituted tropanes. The most important fragment is [ $\left.\mathrm{M}-\mathrm{X}-\mathrm{COO}-\mathrm{CH}=\mathrm{CH}_{2}\right]^{+}$after expulsion of the ethylene bridge $\mathrm{C}-6-\mathrm{C}-7$ including its substituent; this allows a prediction of the substituents' positions in 3,6/7-disubstituted tropanes [7, 8]. Regarding 1, there are two possible key ions: in case of acetylation in position $7 \mathrm{~m} / \mathrm{z} 195$ or in case of acetylation in position $2 \mathrm{~m} / \mathrm{z} 155$. As there is only a veritable peak at $\mathrm{m} / \mathrm{z} 195$, 1 has to be acetylated in position 7 of the tropane.

The relative stereochemistry of 1 was deduced from characteristic coupling constants: H-7 showed a doublet-doublet with coupling constants of 3.4 Hz and 7.9 Hz that can also be observed in the $7 \beta$-substituted schizanthines C-E [9]. This corresponds with the experience that, for steric reasons, bridge substituents usually are exo-orientated. $\mathrm{H}-2$ showed a trans-diaxial coupling constant $J=10 \mathrm{~Hz}$ which is - according to [10] and [11] specific for $\alpha$-orientated substituents at $\mathrm{C}-2$. These conclusions were also confirmed by NOE measurements: $\mathrm{H}-2$ ( $\delta_{\mathrm{H}} 5.02$ ) showed correlations to $\mathrm{H}-1\left(\delta_{H} 3.55\right)$, to the equatorial $\mathrm{H}-3 \mathrm{e}\left(\delta_{H} 1.98\right)$ and to the axial $\mathrm{H}-4 \mathrm{a}\left(\delta_{H} 1.89\right)$ which is only possible if $\mathrm{H}-4 \mathrm{a}$ and $\mathrm{H}-2$ are both axial [11]. H-7 ( $\delta_{H} 4.61$ ) was correlated to $\mathrm{H}-1\left(\delta_{H} 3.55\right)$ and - only enabled by its endo-position - to the axial $\mathrm{H}-3 \mathrm{a}\left(\delta_{H} 1.49\right)$ and $\mathrm{H}-6 \mathrm{n}\left(\delta_{H} 2.36\right)$.

Thus, 1 (ipvelutine) was identified as $7 \beta$-acetoxy- $2 \alpha$-(tigloyloxy)tropane.


Fig. 1. Structure of ipvelutine $[7 \beta$-acetoxy-2 $\alpha$-(tigloyloxy)tropane], main alkaloid from the vegetative parts of Ipomoea velutina R. Br.

In the vegetative parts and/or roots, eight minor compounds related to ipvelutine could be detected by GC-MS analysis. They were identified by their fragmentation patterns; characteristic base peaks of those 2,7-disubstituted tropanes are $\mathrm{m} / \mathrm{z} 95$ and $\mathrm{m} / \mathrm{z} 82$ or $\mathrm{m} / \mathrm{z} 81$ together with a prominent peak at $\mathrm{m} / \mathrm{z} 156$, and of their nortropane derivatives $\mathrm{m} / \mathrm{z}$ 125 and $m / z 81$ including a half-maximal peak at $m / z 108$. An additional result of the systematic GC-MS screening is the detection of ipvelutine (appearing as deacetylated derivative in GC-MS analysis) in vegetative parts of Convolvulus graminetinus, $C$. sagitattus, and Ipomoea abrupta. Both Convolvulus species afforded similar structures, as well, and, additionally, the corresponding nortropanes in the roots. Ipvelutine-related substances were also found in Ipomoea asarifolia and I. plebeia. The mass fragmentation patterns obtained by GC-MS analysis show that these variations include differences in the stereostructure at C-2 or/and C-7, alternation of the position of the substituents, methylbutyric and hydroxymethylbutyric acid as diverging acyl components, change of the
bridge substituents' position from C-7 to C-6 and a hydroxy group as additional substituent (for details see [12]).

2,7-Dihydroxynortropane showing the same substitution pattern as ipvelutine is also synthesized by root cultures of Calystegia sepium (Solanaceae). Incorporation experiments with ${ }^{15} \mathrm{~N}$-labelled 3 -tropanone revealed that, unless 2,7-dihydroxynortropane derives the regular tropane alkaloid pathway, it is not an intermediate in calystegine biosynthesis, but can be seen as a by-product [5].

From the pharmacological point of view, the finding of ipvelutine and derivatives is of interest since they show structural similarity to bao gong teng A [13] obtained from the vegetative parts of Erycibe obtusifolia (Convolvulaceae). Bao gong teng A is characterized by strong miotic properties and therefore used as an antiglaucoma agent in medicinal products. This pharmacological effect is contradictory to that of atropine/hyoscyamine having significance as a mydriatic in ophthalmology and being one of the most commonly used tropanes of natural origin.

## Experimental

## General procedures

${ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$-COSY spectra were obtained on a Bruker AMX 400 MHz , HSQC and HMBC spectra on a Bruker DRX 500 MHz (TMS as internal standard). EIMS and HREIMS were recorded on a Varian MAT 711 ( 80 eV ), FABMS on a Varian MAT CH ${ }_{5}$ DF. The GC-MS system consisted of a Fisons GC 8060 coupled to a quadrupole mass spectrometer Fisons MD 800c.

## Plant material

Roots and vegetative parts of Ipomoea velutina R. Br. grown from seeds collected in the wild at Florence Falls, Litchfield National Park, Northern Territory/Australia, were harvested in the greenhouse of the Institut für Pharmazie, Freie Universität Berlin. A voucher specimen is deposited at the herbarium of the Berlin-Dahlem Botanical Garden Botanical Museum (BGBM), Freie Universität Berlin, Germany.

## Extraction and isolation of ipvelutine

235 g dried and ground vegetative parts of Ipomoea velutina were extracted 4 h with 3 L MeOH three times and once with a mixture of 2.4 L MeOH and $600 \mathrm{~mL} 2 \%$ aqueous tartaric acid. After evaporation of the $\mathrm{MeOH}\left(50^{\circ} \mathrm{C}\right.$ i. V.), the residue was redissolved in $600 \mathrm{~mL} 2 \%$ aqueous tartaric acid and extracted with petrol ether, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, and EtOAc, respectively ( $3 \times 500 \mathrm{~mL}$ each). Then, the aqueous layer was alkalinized ( pH 10 ) with aqueous $\mathrm{NH}_{3}$ (25\%) and extracted with $4 \times 500 \mathrm{mLCH} \mathrm{Cl}_{2}$. The united alkaline $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ fractions gave 172 mg crude alkaloid fraction which was dissolved in $50 \mathrm{~mL} 2 \%$ aqueous tartaric acid again and extracted with petrol ether, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, and EtOAc ( $3 \times 50 \mathrm{~mL}$ each ). After addition of aqueous $\mathrm{NH}_{3}(\mathrm{pH} 10)$, the aqueous layer was extracted with $4 \times 50 \mathrm{~mL}$ $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. After drying over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and evaporation of $\mathrm{CH}_{2} \mathrm{Cl}_{2}\left(40^{\circ} \mathrm{C}\right.$ i. V.), the alkaline fractions were united and 10 mg ipvelutine were gained ( $81 \%$ purity according to NMR spectra).
$7 \beta$-Acetoxy-2 $\alpha$-(tigloyloxy)tropane [(1S,2S,5R,7R)-7-(acetyloxy)-8-methyl-8-aza-bicyclo[3.2.1]oct-2-yl (2E)-2-methylbut-2-enoate, ipvelutine, 1]
Yellow oil. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ ( 400 MHz , MeOD): see Table 1. ${ }^{13} \mathrm{C}-\mathrm{NMR}$ (100.6 MHz, MeOD): see Table 1. MS (EI, $80 \mathrm{eV}, 110^{\circ} \mathrm{C}$ ): $\mathrm{m} / \mathrm{z}(\%)=281$ (2) [M] ${ }^{+}, 239$ (83), 195 (7), 156 (100), 142 (60), 140 (35), 112 (11), 98 (46), 96 (84), 95 (91), 94 (50), 85 (41), 84 (31), 83 (27), 55 (22), 43 (20). (+)-FAB MS (80 eV): $m / z=282[M+H]^{+}$. HR MS (80 eV): $m / z=281.16256$ (calcd. 281.16271 for $\mathrm{C}_{15} \mathrm{H}_{23} \mathrm{NO}_{4}$ ), 239.15283 (calcd. 239.15214 for $\mathrm{C}_{13} \mathrm{H}_{21} \mathrm{NO}_{3}$ ), 156.10254 (calcd. 156.10245 for $\mathrm{C}_{8} \mathrm{H}_{14} \mathrm{NO}_{2}{ }^{+}$), 142.08678 (calcd. 142.08681 for $\mathrm{C}_{7} \mathrm{H}_{12} \mathrm{NO}_{2}{ }^{+}$), 140.10749 (calcd. 140.10754 for $\mathrm{C}_{8} \mathrm{H}_{14} \mathrm{NO}^{+}$), 98.062524 (calcd. 98.06059 for $\mathrm{C}_{5} \mathrm{H}_{8} \mathrm{NO}^{+}$), 95.072728 (calcd. 95.073499 for $\mathrm{C}_{6} \mathrm{H}_{9} \mathrm{~N}$ ).

## GC-MS analysis

Ground plant parts ( 50 g ) were extracted three times with 500 mL MeOH (80\%). After evaporation the residue was dissolved in $2 \%$ aqueous tartaric acid and extracted with petrol ether, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, and EtOAc. The aqueous layer was alkalinized and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. To purify the extracts obtained, this procedure was repeated with corresponding smaller amounts of the solvents. The resulting extracts were subjected to GC-MS analysis. Samples were injected at $240^{\circ} \mathrm{C}$ (split 1:20) and separated on a DB-1 column ( $0.32 \mathrm{~mm} x$ $30 \mathrm{~m}, \mathrm{~J} \& \mathrm{~W}$ Scientific, California) by raising temperature from $70^{\circ} \mathrm{C}$ to $300^{\circ} \mathrm{C}$ at $6^{\circ} \mathrm{C} / \mathrm{min}$. Helium was used as carrier gas. Retention indices (RI): Kovats indices [14] were calculated in relation to a set of co-injected hydrocarbons.

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## Authors' Statement

## Competing Interests

The authors declare no conflict of interest.

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