

5 Appendices

5.1 List of Abbreviations

Chemicals:

DTT	Dithiothreitol
NAD ⁺	Nicotinamide adenine dinucleotide
PCMBS	p-chloromercuribenzene sulfonic acid
PEG	Polyethyleneglycol
PIP	di- μ -iodobis(ethylenediamine)diplatinum(II)nitrate
TCEP	tris-(2-carboxyethyl)phosphine

Datasets:

Nat-2	native protein
Malmn1	native protein with maltose, NAD ⁺ , Mn ²⁺
Ag12	C25S/C174S double mutant with maltose, NAD ⁺ , Mn ²⁺
Ag18	native protein with maltose, Co ²⁺

Crystallographic terms:

\AA	Angstrom, 0.1nm
BESSY II	Berliner Elektronenspeicherring - Gesellschaft für Synchrotronstrahlung m.b.H. II
B-factor	temperature factor
CAZY	Carbohydrate Active Enzyme Database
CSW	PDB abbreviation for cysteine-sulfenic acid

Appendices

Cys-SOH	Cysteine-sulfenic acid
Cys-SO ₂ H	Cysteine-sulfinic acid
DESY	Deutsches Elektronen Synchrotron
EMBL	European Molecular Biology Laboratory
GH4	Glycosyl Hydrolase family 4
MAD	Multiple wavelength Anomalous Dispersion
MIR	Multiple Isomorphous Replacement
MIRAS	Multiple Isomorphous Replacement with Anomolous Scattering
PDB	Protein Data Bank
rmsd	root mean square deviation

5.2 Theory of X-ray Crystallography

5.2.1 Diffraction of Waves: Bragg's Law

An X-ray is an electromagnetic wave with a wavelength of around 1 Å. As with all waves, it may be characterized by its wavelength, amplitude and phase (Figure 5.1, A). Diffraction of a wave occurs when the wave is scattered into directions other than that of the original beam without changing the wavelength. When an X-ray hits an electron, it sets the electron oscillating with the X-ray frequency, which may then radiate an X-ray photon of the same wavelength in a random direction as it returns to its unexcited state.

According to Bragg, the diffraction of X-rays by a crystal lattice can be described as reflection from a series of lattice planes. When an incoming wave (incident beam) hits a lattice plane under the angle θ , it will be reflected (diffracted) to produce an outgoing wave (diffracted beam) that will have the same wavelength as the incident beam, however, the phase may differ. This diffraction is shown for beam A in Figure 5.1, B. The same diffraction will occur for all planes in the series, however the extra distance traveled, described by $2d\sin\theta$, will result in a phase shift of the diffracted beam. Positive interference will only occur if the extra distance the second beam travels (d) is equal to an integer number of wavelengths ($n\lambda$), as then the scattered beam is in phase with the first. This is described by Bragg's Law:

$$n\lambda = 2d\sin\theta$$

A set of parallel planes in a lattice is described by the indices h, k, l . These represent the fractional distance along each of the unit cell axes a, b, c which separate each plane in the series (Figure 5.1, C). Therefore each X-ray beam reflected from a set of planes can also be described by a set of indices h, k, l .

Appendices

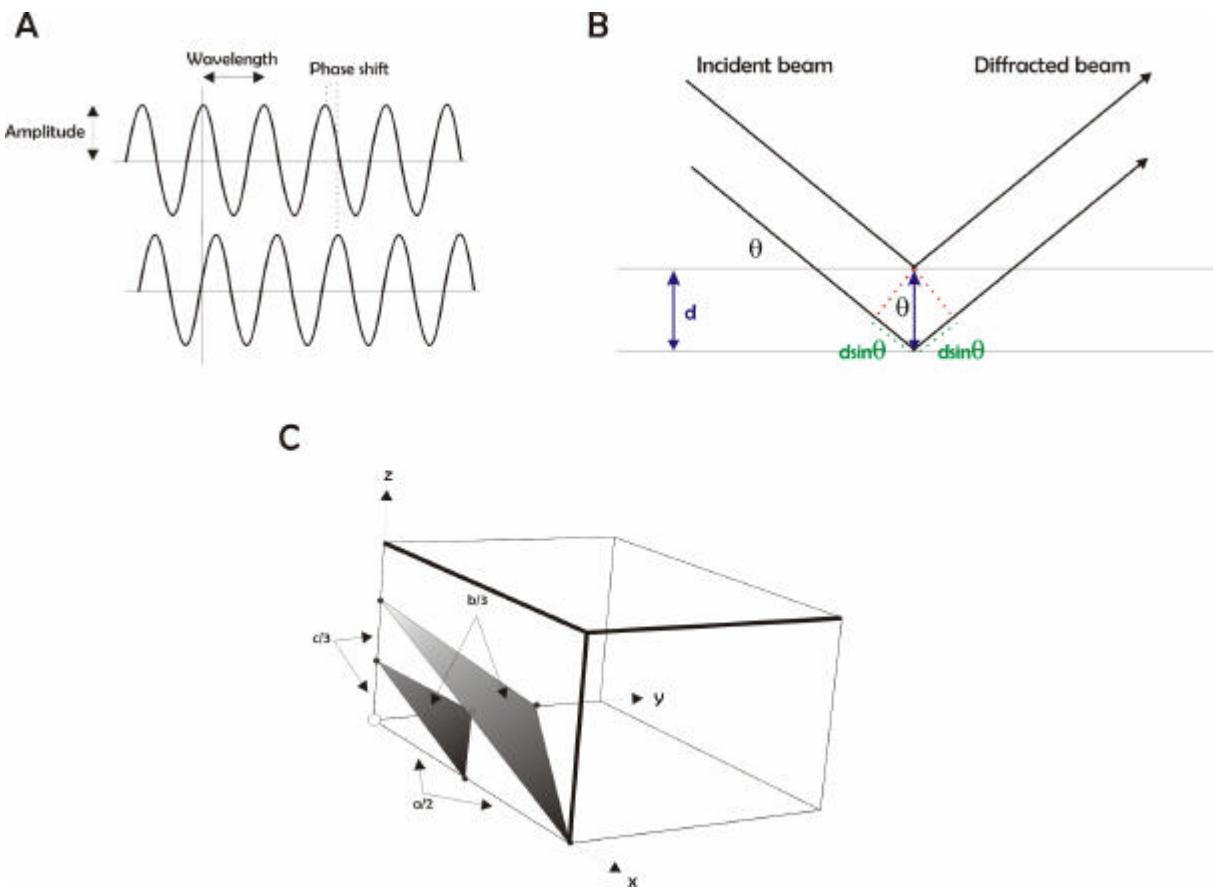


Figure 5.1 Diffraction of waves from planes

A. Wavelength and amplitude of a sinusoidal wave. Also shown is a second sinusoidal wave of the same wavelength and amplitude, but with a phase shift compared to the top wave. **B.** Bragg's Law. Shown are two planes of a crystal lattice with a distance of d . The difference between the distance traveled by the upper beam and the lower beam is shown as a green dotted line. **C.** Three-dimensional view of a unit cell, showing an example of the planes characterised by the h, k, l indices 2, 3, 3.

5.2.2 Reciprocal Space and the Ewald Sphere

In 1921, Ewald introduced a geometric construction which illustrates the relationship between the orientation of the crystal in the beam and the resulting X-ray reflections (Figure 5.2). A sphere is drawn around the crystal in position X. The X-ray beam enters this sphere in point I, and hits the crystal under the angle θ . The reciprocal lattice is a lattice with the same symmetry elements as the real crystal lattice, but with unit cell dimensions a^*, b^*, c^* , which have the following relationship to the unit cell dimensions a, b, c of the real lattice:

$$a^* = \frac{1}{a}, \quad b^* = \frac{1}{b}, \quad c^* = \frac{1}{c}$$

Appendices

At the point where the X-ray beam leaves the sphere, the origin (O) of a reciprocal lattice is placed. Based on this construction, for a given angle θ , only those lattice points of the reciprocal lattice fulfill Bragg's law which fall onto the circumference of the sphere (here point P). These result in a reflection. For a complete diffraction data set the crystal has to be rotated in such a way that each lattice point of the reciprocal lattice passes through the Ewalds sphere at least once. For more details, see (Drenth, 1994; McRee, 1999).

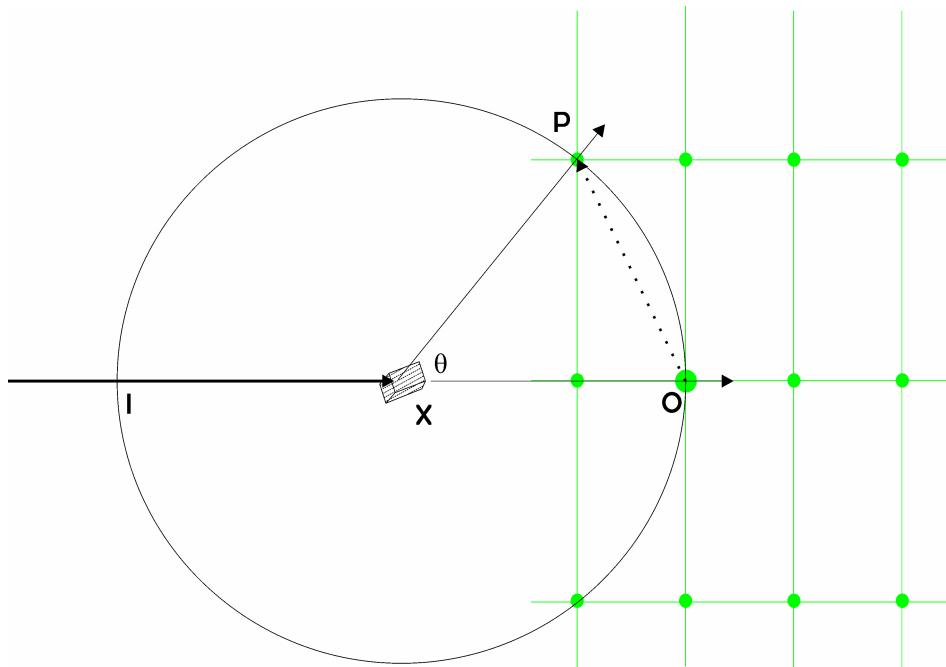


Figure 5.2 Ewald sphere and the reciprocal lattice

5.2.3 The Phase Problem

The diffracted beam is scattered by the electrons in the protein. Therefore the diffraction data in principle contain the information on the location of the electrons in the unit cell. The electron density (?) at a certain position in the unit cell (x, y, z) is a function of the amplitude ($|F|$) and the phase (α) of all diffracted beams:

$$r(xyz) = \frac{1}{V} \sum_{hkl} |F(hkl)| \exp[-2\pi i(hx + ky + lz) + i\alpha(hkl)]$$

Appendices

The amplitudes or structure factors $|F|$ are proportional to the intensities of the reflections and can thus be determined experimentally. However, the phase information is lost during the data collection. This is called the phase problem of crystallography.

5.2.4 Solving the Phase Problem by Isomorphous Replacement

The method of multiple isomorphous replacement (MIR) was first used by Max Perutz in 1953. It relies on the observation that a heavy atom, having a much higher concentration of electrons in a small volume, will scatter X-rays strongly. If a heavy atom can be introduced into a protein crystal via soaking or co-crystallisation, without otherwise changing the protein or the crystal lattice (i.e. the crystal remains isomorphous), the differences in the scattering intensities from this crystal to the native crystal can be exploited to indirectly determine the missing phase angles.

The structure factors from the native crystal ($|F_P|$) and those of the heavy atom-derivatised crystal ($|F_{PH}|$) are scaled together to remove any variation caused by crystal quality. $|F_P|$ is then subtracted from $|F_{PH}|$, which gives the scattering intensities for only the heavy atoms ($|F_H|$). This information is then used for the Patterson synthesis, a fourier synthesis in which the phases are assumed to be zero, the amplitudes are squared and a center of symmetry is added (Drenth, 1994). This will produce a map which gives the vectors between the heavy atoms in real space, however not their actual position. Using these vectors it is possible to determine the positions of the heavy atoms in the unit cell. With these sites, and the structure factors, the Fourier transformation can now be used to determine the unknown phases for these heavy atoms. From the heavy atom phases, the phases of the protein can be derived, and the initial electron density map of the protein can be calculated.

At this step there are still two possible solutions for every phase angle, only one of which is correct. This is the major limitation to structure determination via Single Isomorphous Replacement (SIR). The ambiguity can only be resolved by adding further information. In the Multiple Isomorphous Replacement (MIR) method, this is provided by further derivatives. Another method is to additionally use the anomalous signal produced by heavy atoms incorporated into the crystal (Single or Multiple Isomorphous Replacement with Anomalous Signal, SIRAS / MIRAS). Alternatively, data from a single crystal with one or more heavy atoms incorporated can be collected at several wavelengths (Multiple Wavelength Anomalous Dispersion, MAD). This has the advantage that non-isomorphism between different crystals is avoided.

Appendices

5.2.5 Crystallographic Refinement

The nature of the data used to build the protein model, from data collection to phase solution to map calculation, is inherently full of error. To correct the model for this error, a crystallographic refinement is carried out.

In this, small shifts in the coordinates (bonds and angles) are introduced and the resulting energy values of the molecule are calculated. The aim is to identify the conformation with the lowest energy and the best fit to the experimental data (McRee, 1999). This is equivalent to finding the minimum of a target function E with:

$$E = E_{\text{chem}} + w_{\text{xray}} E_{\text{xray}}$$

where E_{chem} is the sum of the energy for all covalent and non-covalent interactions, including bond lengths and angles, torsion angles, non-bonded contacts, planar groups and chiral volumes. E_{xray} describes the differences between the observed structure factors, F_o , and those which can be calculated from the model, F_c :

$$E_{\text{xray}} = \sum_{hkl} \left[|F_o(h)| - k |F_c(h)| \right]^2$$

where h is comprised of all the h,k,l indices of all reflections. w_{xray} is a weighting term which allows fine tuning of the influence of the X-ray term on the overall target function. A critical aspect of the refinement procedure is that the number of refined parameters (for each protein atom the position x, y, z and the B-factor) may not exceed the number of observations, that is the number of reflections collected from a crystal. The number of parameters can be reduced indirectly by restraining the stereochemistry to values derived from small molecule crystallography (Engh and Huber, 1991). The outcome of the refinement procedure therefore is an atomic model of the protein with optimised stereochemistry and a minimised overall energy which best fits the X-ray data collected.

The progress of the refinement is judged by the crystallographic R-factor, R_{cryst} , which describes the deviation between the measured structure factors F_o and calculated F_c .

Appendices

$$R_{\text{cryst}} = \frac{\sum_{hkl} |F_o(h) - k|F_c(h)|}{\sum_{hkl} |F_o(h)|}$$

As this factor is similar to the X-ray term, it may be artificially lowered when too many parameters are refined at the same time (over-fitting). A reference factor, termed R_{free} , was introduced by Brünger, where a subset of the reflections (usually 5-10% of the data collected) are excluded from refinement (test set) (Brünger, 1993). The R_{free} is then calculated as for R_{cryst} but only against this test set.

5.3 Theory of SAXS

5.3.1 SAXS in an Ideal Solution

SAXS of biological macromolecules in solution follows the same rules that govern the scattering of X-rays by electrons in solid matter. As the distances involved in these proteins are much greater than interatomic distances, the scattering particles are described with a continuous electron density. In the following discussion the electron density at a single point, r , is represented by $p(r)$ and the characteristic vector of the incident beam as s_0 .

An ideal solution is one in which the particles within do not interact with each other and whose thermodynamics quantities follow a linear dependence on concentration. This is approximated in SAXS by the use of extremely dilute solutions and the measurement of a range of concentrations. The monodispersity of a solution describes the identity of the molecules within and holds true for particles with a defined shape such as native proteins. In several cases, such as denatured proteins, the molecules may be monodisperse in size, but show a large degree of polydispersity in shape. It is also possible to have polydispersity in size but monodispersity in shape with particles such as micelles, and polydispersity for both can be observed for aggregations. In an ideal monodisperse solution all particles scatter independently and the measured intensity is merely the sum of the intensities from each individual molecule:

$$I(s) = N i_1(s)$$

$I(s)$ = experimental scattering intensity

N = number of particles in sample

$i_1(s)$ = scattered intensity of the particle

It is possible to directly determine the size of the particle by knowing the intensity at the zero angle ($s=0$) with the formula:

$$I(0)/c = \frac{N_A M}{\mu^2} (1 - \rho_0 \psi)^2$$

$I(0)$ = experimental scattering intensity at angle 0

c = concentration

N_A = Avagadro's constant

M = molecular weight

μ = M/m , ratio of M to the number of electrons

ρ_0 = electron density distribution of particle at angle 0

ψ = electronic partial specific volume of particle

Appendices

Therefore the intensity at zero angle, scaled by the concentration, is proportional to the molecular weight and if ρ_0 , ψ and the intensity of the incident beam are known, the intensity at the origin will give the molecular weight.

5.3.2 Radius of Gyration

The movement of a particle within a solution follows Brownian motion. An ideal suspension of many particles would therefore assume all particles to be at all positions and in all orientations. This isotropic intensity distribution is proportional to the scattering for a single particle averaged over all orientations. Close to the origin (angle = 0), the scattering pattern of any single particle can be approximated by a Gaussian, the width of which is proportional to the square of the radius of gyration of the particle: the Guinier relation (Guinier and Fournet, 1955).

$$R_g^2 = \frac{\int_V \Delta\rho(\mathbf{r}) r^2 dV_r}{\int_V \Delta\rho(\mathbf{r}) r dV_r}$$

Here, the origin of the vector is taken to be the centre of mass of the particle, and the radius of gyration (R_g) describes its sphericity. The relation of the radius of gyration to the scattering intensity close to the origin is described by:

$$I(s) \cong I(0) \exp\left(-\frac{4\pi^2}{3} R_g^2 s^2\right)$$

By graphing the natural log of $I(s)$ against s^2 , a linearised representation is created, as Guinear plot (Figure 5.3). From this plot, linear regression yields the radius of gyration from the slope, together with the intensity at the origin. However, a Guinear approximation does rely on an ideal monodisperse solution.

Appendices

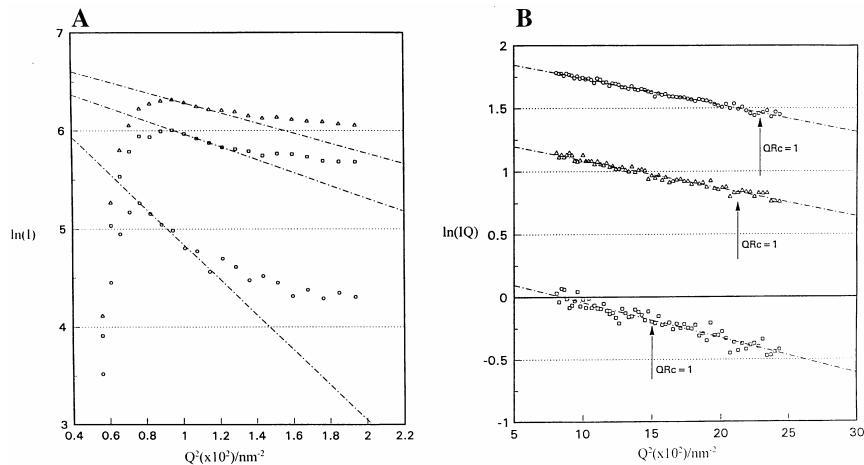


Figure 5.3 Plots of the protein K-gliadin

A. Guinier (R_g) plots for K-gliadin in 70% (v/v) aq. ethanol at concentrations of 9.3 (a), 4.7 (*), 2.3 (E) and 1.3 (O) mg/ml. Limits for the Guinier approximation to be valid ($QR_g \approx 1$) are shown. **B.** Cross-section (R_c) plots at concentrations of 9.3 (a), 4.7 (O), 2.3 (E) and 1.3 (*) mg/ml. Limits for the cross-section approximation to be valid ($QR_c \approx 1$) are shown. Taken from (Thomson et al., 1999).

5.3.3 SAXS Data Processing

The scattering intensity from a single particle in an ideal solution cannot be measured directly, rather a set of intensities ($I_{exp}(s_i)$ where $I=1\dots N$) in a restricted angular range $s_{min} < s < s_{max}$ is measured. This set will contain statistical errors and inaccuracies produced by the beam divergence, detector resolution etc, and so the task of data processing is restoring the $I(s)$ from $I_{exp}(s_i)$. For an ideal solution, $I(s)$ is related to the pair distribution function of the particle $\rho(\mathbf{r})$ by the Fourier transform

$$I(s) = 4\pi \int_0^{D_{max}} \rho(r) \frac{\sin 2\pi s r}{2\pi s r} dr$$

D_{max} = maximum particle diameter

The function $\rho(\mathbf{r})$ contains the same information as $I(s)$ and the data processing can be done indirectly by restoring the $\rho(\mathbf{r})$. By providing the $\rho(\mathbf{r})$, the function $I(s)$ is automatically extrapolated beyond the measured range of scattering vectors back to the origin. By using the entire scattering curve, the system is much less sensitive to small amounts of interactions or aggregations than the Guinier relation, and, like the latter, the effects of these interactions are restricted to the innermost part of the scattering pattern.

Appendices

The program GNOM performs this indirect transform, identifying the multiplier which compensates between the quality of fit to the data and the smoothness of the $p(\mathbf{r})$ function. The program visually compares solutions for different multiplier values automatically, finding the optimal solution or indicating that there may be problems in the assumptions (such as particle size, D_{\max}) used.

5.3.4 Shape Determination

The analysis of a protein with known structure (ie. from crystallography or NMR results) is performed by the program CRYSTAL (Svergun et al., 1995), which, as well as predicting the scattering curve produced by the protein, also takes into account the surrounding hydration shell and its effect on scattering.

The *ab initio* shape determination may take two approaches: the first is applied by the program SASHA and employs spherical harmonics to describe the particle shape, thereby making it possible to directly restore the shape from the scattering data. (Kozin et al., 1997). The second approach utilised by the programs DAMMIN and GASBOR, is based on a trial and error analysis of scattering patterns produced by models filled with spheres (dummy atoms) or, in the latter case, dummy residues. Examples of this approach are shown in Figure 5.4.

5.3.5 Instrumentation

The X-ray beam originating from a synchrotron source passes through a monochromator which produces a monochromatic beam. The vertical beam displacement reduces background radiation from the synchrotron, and reflection at a glancing angle from the toroidal mirror eliminates harmonics and focuses in the two directions normal to the beam. The pairs of slits eliminate parasitic scattering around the monochromatic beam, essential to reach the small angles required by SAXS. The camera contains another three pairs of slits, also to remove parasitic scattering, and defines the minimum observable scattering angle. The sample is fixed in a temperature controlled quartz capillary, but the sample-to-detector distance can be adjusted by moving the detector within the range of 0.5 and 10m.

Appendices

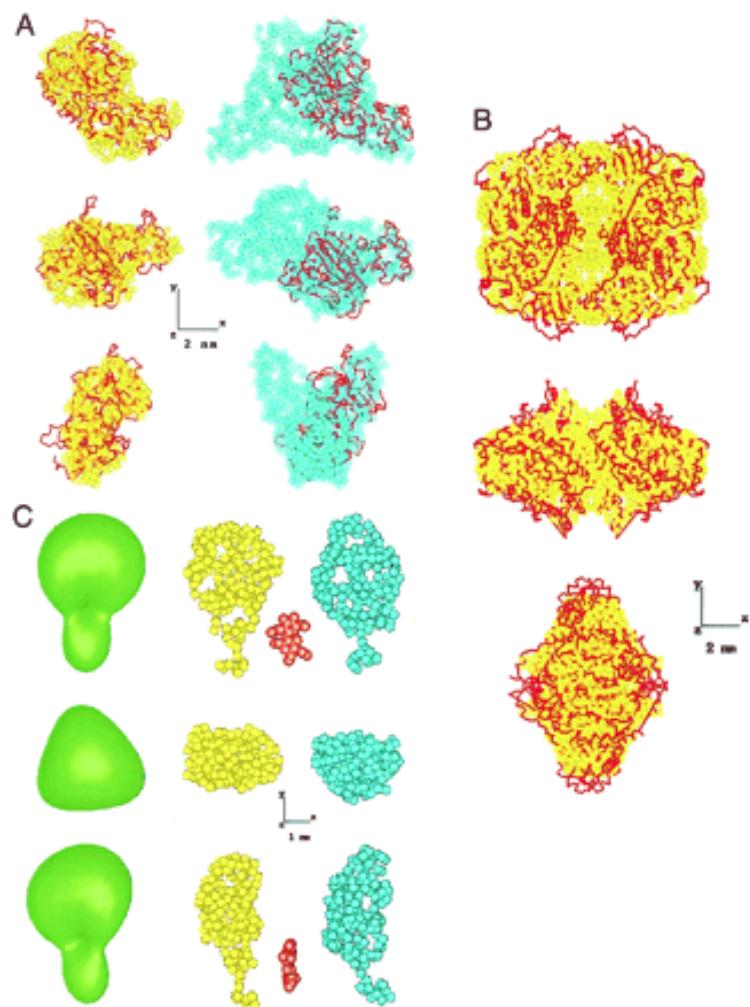


Figure 5.4 *Ab initio* Shape Determination with SAXS

A. Comparison of the atomic model of monomeric hexokinase (*red*) and the dummy residue (DR) models (*yellow*) of the monomeric (left) and dimeric (right) hexokinase. **B.** Atomic model of tetrameric yeast pyruvate decarboxylase superimposed with the DR model. **C.** *Ab initio* models of chitin binding protein: left, low resolution envelope produced by SASHA, middle and right, two most different DR models. The α -chitin substrate is shown in *red*. From (Svergun et al., 2001).

Appendices

5.4 CNS Files for Maltose

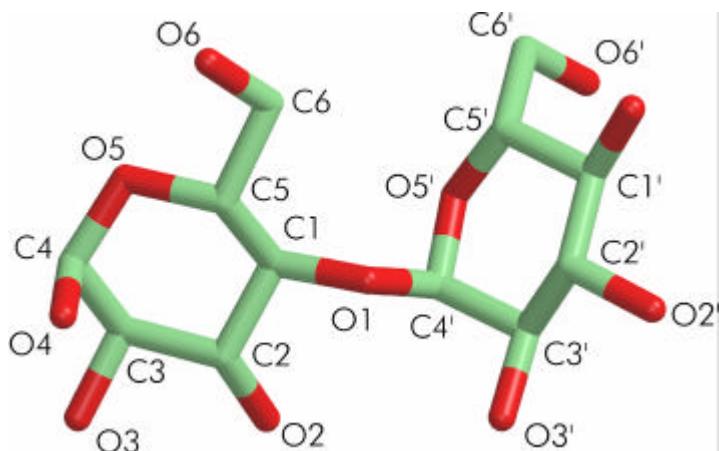


Figure 5.5 Schematic diagram of maltose

Apostrophes in atom labels follow standard PDB nomenclature.

5.4.1 CNS Parameter File for Maltose

BOND	CM1	CM2	1000.0	1.520	
BOND	CM1	OM7	1000.0	1.412	
BOND	CM1	OM11	1000.0	1.401	
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BOND	CM3	CM4	1000.0	1.529	
BOND	CM3	OM9	1000.0	1.431	
BOND	CM4	CM5	1000.0	1.534	
BOND	CM4	OM10	1000.0	1.443	
BOND	CM5	CM6	1000.0	1.537	
BOND	CM5	OM11	1000.0	1.437	
BOND	CM6	OM12	1000.0	1.423	
BOND	OM7	CM16	1000.0	1.438	
BOND	CM13	CM14	1000.0	1.507	
BOND	CM13	OM22	1000.0	1.416	
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BOND	CM17	CM18	1000.0	1.539	
BOND	CM17	OM22	1000.0	1.458	
BOND	CM18	OM23	1000.0	1.440	
BOND	CM13	OM19	1000.0	1.412	
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ANGLE	CM5	CM4	OM10	500.0	107.04
ANGLE	CM4	CM5	CM6	500.0	114.21

Appendices

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ANGLEe	CM16	CM17	OM22	500.0	116.29
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DIHEDral	OM7	CM1	OM11	CM5	750.0 0 90.00
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DIHEDral	CM1	CM2	CM3	OM9	750.0 0 180.00
DIHEDral	OM8	CM2	CM3	CM4	750.0 0 180.00
DIHEDral	OM8	CM2	CM3	OM9	750.0 0 60.00
DIHEDral	CM2	CM3	CM4	CM5	750.0 0 60.00
DIHEDral	CM2	CM3	CM4	OM10	750.0 0 180.00
DIHEDral	OM9	CM3	CM4	CM5	750.0 0 180.00
DIHEDral	OM9	CM3	CM4	OM10	750.0 0 -60.00
DIHEDral	OM22	CM13	CM14	CM15	750.0 0 60.00
DIHEDral	CM13	CM14	CM15	CM16	750.0 0 -60.00
DIHEDral	CM13	CM14	CM15	OM21	750.0 0 180.00
DIHEDral	OM20	CM14	CM15	CM16	750.0 0 180.00
DIHEDral	OM20	CM14	CM15	OM21	750.0 0 60.00
DIHEDral	CM14	CM15	CM16	CM17	750.0 0 60.00
DIHEDral	OM21	CM15	CM16	CM17	750.0 0 180.00
DIHEDral	OM7	CM16	CM17	OM22	750.0 0 180.00
DIHEDral	CM15	CM16	CM17	CM18	750.0 0 180.00
DIHEDral	CM15	CM16	CM17	OM22	750.0 0 -60.00
DIHEDral	CM16	CM17	CM18	OM23	750.0 0 180.00
DIHEDral	OM22	CM17	CM18	OM23	750.0 0 60.00
DIHEDral	CM16	CM17	OM22	CM13	750.0 0 60.00
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IMPRoper	CM5	CM4	CM6	OM11	750.0 0 -35.000
IMPRoper	CM14	CM13	CM15	OM20	750.0 0 -35.000
IMPRoper	CM15	CM14	CM16	OM21	750.0 0 35.000
IMPRoper	CM16	OM7	CM15	CM17	750.0 0 -35.000
IMPRoper	CM17	CM16	CM18	OM22	750.0 0 -35.000
IMPRoper	CM13	CM14	OM19	OM22	750.0 0 -35.000
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NONBonded	CM6	0.1200	3.7418	0.1000	3.3854

Appendices

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NONBonded	CM14	0.1200	3.7418	0.1000	3.3854
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NONBonded	CM18	0.1200	3.7418	0.1000	3.3854
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NONBonded	OM21	0.1591	2.8509	0.1591	2.8509
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NONBonded	OM23	0.1591	2.8509	0.1591	2.8509

5.4.2 CNS Topology File for Maltose

MASS	CM1	13.01900			
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MASS	CM3	13.01900			
MASS	CM4	13.01900			
MASS	CM5	13.01900			
MASS	CM6	14.02700			
MASS	OM7	15.99900			
MASS	OM8	17.00700			
MASS	OM9	17.00700			
MASS	OM10	17.00700			
MASS	OM11	15.99900			
MASS	OM12	17.00700			
MASS	CM13	14.02700			
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MASS	CM17	13.01900			
MASS	CM18	14.02700			
MASS	OM19	17.00700			
MASS	OM20	17.00700			
MASS	OM21	15.99900			
MASS	OM22	17.00700			
ATOM	C1	TYPE CM1	CHARge	0.0	END
ATOM	C2	TYPE CM2	CHARge	0.0	END
ATOM	C3	TYPE CM3	CHARge	0.0	END
ATOM	C4	TYPE CM4	CHARge	0.0	END
ATOM	C5	TYPE CM5	CHARge	0.0	END
ATOM	C6	TYPE CM6	CHARge	0.0	END
ATOM	O1	TYPE OM7	CHARge	0.0	END
ATOM	O2	TYPE OM8	CHARge	0.0	END
ATOM	O3	TYPE OM9	CHARge	0.0	END
ATOM	O4	TYPE OM10	CHARge	0.0	END
ATOM	O5	TYPE OM11	CHARge	0.0	END
ATOM	O6	TYPE OM12	CHARge	0.0	END
ATOM	C1'	TYPE CM13	CHARge	0.0	END
ATOM	C2'	TYPE CM14	CHARge	0.0	END
ATOM	C3'	TYPE CM15	CHARge	0.0	END
ATOM	C4'	TYPE CM16	CHARge	0.0	END
ATOM	C5'	TYPE CM17	CHARge	0.0	END
ATOM	C6'	TYPE CM18	CHARge	0.0	END
ATOM	O1'	TYPE OM19	CHARge	0.0	END
ATOM	O2'	TYPE OM20	CHARge	0.0	END
ATOM	O3'	TYPE OM21	CHARge	0.0	END
ATOM	O5'	TYPE OM22	CHARge	0.0	END

Appendices

```
ATOM O6' TYPE OM23 CHARge 0.0 END  
BOND C1 C2 BOND C1 O1 BOND C1 O5 BOND C2 C3  
BOND C2 O2 BOND C3 C4 BOND C3 O3 BOND C4 C5  
BOND C4 O4 BOND C5 C6 BOND C5 O5 BOND C6 O6  
BOND O1 C4' BOND C1' C2' BOND C1' O5' BOND C2' C3'  
BOND C2' O2' BOND C3' C4' BOND C3' O3' BOND C4' C5'  
BOND C5' C6' BOND C5' O5' BOND C6' O6' BOND C1' O1'  
  
DIHEdral O5 C1 C2 O2  
DIHEdral C1 C2 C3 O3  
DIHEdral O2 C2 C3 C4  
DIHEdral C2 C3 C4 O4  
DIHEdral O3 C3 C4 C5  
DIHEdral C1' C2' C3' O3'  
DIHEdral O2' C2' C3' C4'  
DIHEdral O3' C3' C4' C5'  
DIHEdral O1 C4' C5' O5'  
DIHEdral C3' C4' C5' C6'  
DIHEdral C4' C5' C6' O6'  
DIHEdral C6' C5' O5' C1'  
  
IMPRoper C1 C2 O1 O5  
IMPRoper C2 C1 C3 O2  
IMPRoper C3 C2 C4 O3  
IMPRoper C4 C3 C5 O4  
IMPRoper C5 C4 C6 O5  
IMPRoper C1' C2' O1' O5'  
IMPRoper C2' C1' C3' O2'  
IMPRoper C3' C2' C4' O3'  
IMPRoper C4' O1 C3' C5'  
IMPRoper C5' C4' C6' O5'  
  
ACCEptor O1 C1  
ACCEptor O2 C2  
ACCEptor O3 C3  
ACCEptor O4 C4  
ACCEptor O5 C1  
ACCEptor O6 C6  
ACCEptor O2' C2'  
ACCEptor O3' C3'  
ACCEptor O5' C1'  
ACCEptor O6' C6'  
ACCEptor O1' C1'
```

Appendices

5.5 CNS files for NAD⁺

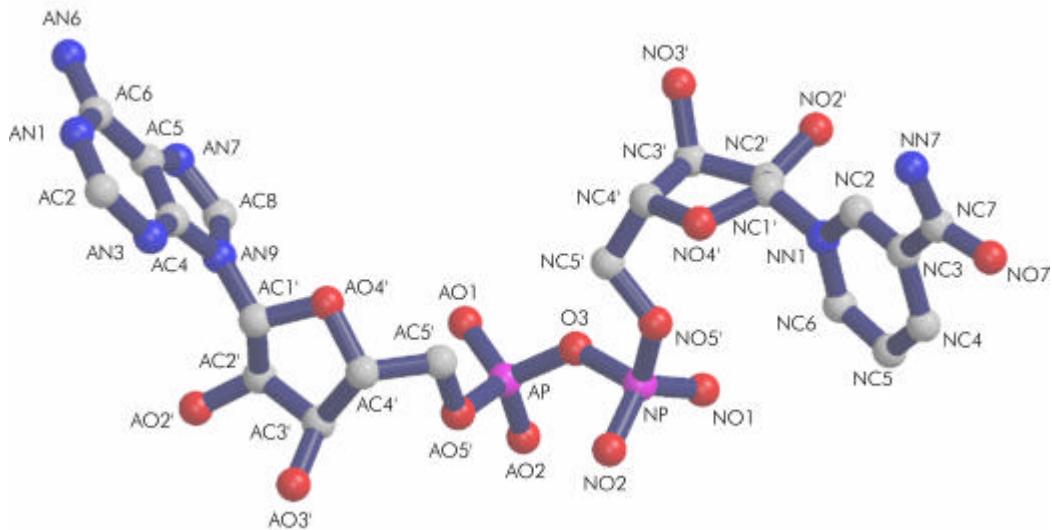


Figure 5.6 Nomenclature of NAD⁺.

Atoms are labelled according to standard nomenclature. The apostrophe on the sugar ring atom labels are replaced by an asterisk in CNS and PDB files, according to CNS and PDB convention.

5.5.1 CNS Parameter File for NAD⁺

BOND	PQ1	OQ2	1000.0	1.482
BOND	PQ1	OQ3	1000.0	1.495
BOND	PQ1	OQ4	1000.0	1.590
BOND	PQ1	OQ23	1000.0	1.594
BOND	OQ4	CQ5	1000.0	1.464
BOND	CQ5	CQ6	1000.0	1.513
BOND	CQ6	OQ7	1000.0	1.482
BOND	CQ6	CQ8	1000.0	1.571
BOND	OQ7	CQ12	1000.0	1.412
BOND	CQ8	OQ9	1000.0	1.458
BOND	CQ8	CQ10	1000.0	1.521
BOND	CQ10	OQ11	1000.0	1.390
BOND	CQ10	CQ12	1000.0	1.540
BOND	CQ12	NQ13	1000.0	1.449
BOND	NQ13	CQ14	1000.0	1.386
BOND	NQ13	CQ22	1000.0	1.402
BOND	CQ14	NQ15	1000.0	1.322
BOND	NQ15	CQ16	1000.0	1.343
BOND	CQ16	CQ17	1000.0	1.459
BOND	CQ16	CQ22	1000.0	1.405
BOND	CQ17	NQ18	1000.0	1.318
BOND	CQ17	NQ19	1000.0	1.369
BOND	NQ19	CQ20	1000.0	1.331
BOND	CQ20	NQ21	1000.0	1.386
BOND	NQ21	CQ22	1000.0	1.382
BOND	OQ23	PQ24	1000.0	1.641
BOND	PQ24	OQ25	1000.0	1.476
BOND	PQ24	OQ26	1000.0	1.487
BOND	PQ24	OQ27	1000.0	1.622
BOND	OQ27	CQ28	1000.0	1.461

Appendices

BOND	CQ28	CQ29	1000.0	1.479	
BOND	CQ29	OQ30	1000.0	1.486	
BOND	CQ29	CQ31	1000.0	1.574	
BOND	OQ30	CQ35	1000.0	1.437	
BOND	CQ31	OQ32	1000.0	1.437	
BOND	CQ31	CQ33	1000.0	1.544	
BOND	CQ33	OQ34	1000.0	1.401	
BOND	CQ33	CQ35	1000.0	1.573	
BOND	CQ35	NQ36	1000.0	1.493	
BOND	NQ36	CQ37	1000.0	1.424	
BOND	NQ36	CQ44	1000.0	1.409	
BOND	CQ37	CQ38	1000.0	1.322	
BOND	CQ38	CQ39	1000.0	1.509	
BOND	CQ38	CQ42	1000.0	1.510	
BOND	CQ39	OQ40	1000.0	1.249	
BOND	CQ39	NQ41	1000.0	1.338	
BOND	CQ42	CQ43	1000.0	1.498	
BOND	CQ43	CQ44	1000.0	1.412	
ANGLE	OQ2	PQ1	OQ3	500.0	109.50
ANGLE	OQ2	PQ1	OQ4	500.0	109.50
ANGLE	OQ2	PQ1	OQ23	500.0	109.50
ANGLE	OQ3	PQ1	OQ4	500.0	109.50
ANGLE	OQ3	PQ1	OQ23	500.0	109.50
ANGLE	OQ4	PQ1	OQ23	500.0	109.50
ANGLE	PQ1	OQ4	CQ5	500.0	109.50
ANGLE	OQ4	CQ5	CQ6	500.0	107.46
ANGLE	CQ5	CQ6	OQ7	500.0	105.98
ANGLE	CQ5	CQ6	CQ8	500.0	115.86
ANGLE	OQ7	CQ6	CQ8	500.0	104.21
ANGLE	CQ6	OQ7	CQ12	500.0	106.74
ANGLE	CQ6	CQ8	OQ9	500.0	110.27
ANGLE	CQ6	CQ8	CQ10	500.0	106.25
ANGLE	OQ9	CQ8	CQ10	500.0	113.70
ANGLE	CQ8	CQ10	OQ11	500.0	117.96
ANGLE	CQ8	CQ10	CQ12	500.0	100.43
ANGLE	OQ11	CQ10	CQ12	500.0	114.32
ANGLE	OQ7	CQ12	CQ10	500.0	106.17
ANGLE	OQ7	CQ12	NQ13	500.0	105.67
ANGLE	CQ10	CQ12	NQ13	500.0	109.94
ANGLE	CQ12	NQ13	CQ14	500.0	132.69
ANGLE	CQ12	NQ13	CQ22	500.0	124.32
ANGLE	CQ14	NQ13	CQ22	500.0	102.96
ANGLE	NQ13	CQ14	NQ15	500.0	113.38
ANGLE	CQ14	NQ15	CQ16	500.0	107.25
ANGLE	NQ15	CQ16	CQ17	500.0	132.75
ANGLE	NQ15	CQ16	CQ22	500.0	108.46
ANGLE	CQ17	CQ16	CQ22	500.0	118.78
ANGLE	CQ16	CQ17	NQ18	500.0	123.00
ANGLE	CQ16	CQ17	NQ19	500.0	114.10
ANGLE	NQ18	CQ17	NQ19	500.0	122.90
ANGLE	CQ17	NQ19	CQ20	500.0	125.44
ANGLE	NQ19	CQ20	NQ21	500.0	123.00
ANGLE	CQ20	NQ21	CQ22	500.0	114.84
ANGLE	NQ13	CQ22	CQ16	500.0	107.94
ANGLE	NQ13	CQ22	NQ21	500.0	128.27
ANGLE	CQ16	CQ22	NQ21	500.0	123.79
ANGLE	PQ1	OQ23	PQ24	500.0	133.63
ANGLE	OQ23	PQ24	OQ25	500.0	108.96
ANGLE	OQ23	PQ24	OQ26	500.0	108.66
ANGLE	OQ23	PQ24	OQ27	500.0	97.94
ANGLE	OQ25	PQ24	OQ26	500.0	121.72
ANGLE	OQ25	PQ24	OQ27	500.0	106.33
ANGLE	OQ26	PQ24	OQ27	500.0	110.61
ANGLE	PQ24	OQ27	CQ28	500.0	119.28
ANGLE	OQ27	CQ28	CQ29	500.0	110.21
ANGLE	CQ28	CQ29	OQ30	500.0	109.29
ANGLE	CQ28	CQ29	CQ31	500.0	114.36

Appendices

ANGLEe	OQ30	CQ29	CQ31	500.0	104.90
ANGLEe	CQ29	OQ30	CQ35	500.0	109.81
ANGLEe	CQ29	CQ31	OQ32	500.0	110.36
ANGLEe	CQ29	CQ31	CQ33	500.0	103.15
ANGLEe	OQ32	CQ31	CQ33	500.0	107.66
ANGLEe	CQ31	CQ33	OQ34	500.0	111.87
ANGLEe	CQ31	CQ33	CQ35	500.0	100.58
ANGLEe	OQ34	CQ33	CQ35	500.0	112.07
ANGLEe	OQ30	CQ35	CQ33	500.0	102.68
ANGLEe	OQ30	CQ35	NQ36	500.0	109.06
ANGLEe	CQ33	CQ35	NQ36	500.0	115.23
ANGLEe	CQ35	NQ36	CQ37	500.0	115.42
ANGLEe	CQ35	NQ36	CQ44	500.0	120.53
ANGLEe	CQ37	NQ36	CQ44	500.0	120.00
ANGLEe	NQ36	CQ37	CQ38	500.0	120.00
ANGLEe	CQ37	CQ38	CQ39	500.0	120.87
ANGLEe	CQ37	CQ38	CQ42	500.0	120.00
ANGLEe	CQ39	CQ38	CQ42	500.0	117.69
ANGLEe	CQ38	CQ39	OQ40	500.0	118.04
ANGLEe	CQ38	CQ39	NQ41	500.0	119.33
ANGLEe	OQ40	CQ39	NQ41	500.0	120.61
ANGLEe	CQ38	CQ42	CQ43	500.0	120.00
ANGLEe	CQ42	CQ43	CQ44	500.0	120.00
ANGLEe	NQ36	CQ44	CQ43	500.0	120.00
DIHEDral	OQ23	PQ1	OQ4	CQ5	750.0 0 90.00
DIHEDral	OQ4	PQ1	OQ23	PQ24	750.0 0 90.00
DIHEDral	OQ4	CQ5	CQ6	OQ7	50.0 3 180.00
DIHEDral	OQ4	CQ5	CQ6	CQ8	750.0 0 -60.00
DIHEDral	OQ7	CQ6	CQ8	CQ10	750.0 0 0.00
DIHEDral	OQ9	CQ8	CQ10	CQ12	750.0 0 90.00
DIHEDral	CQ12	NQ13	CQ14	NQ15	750.0 0 180.00
DIHEDral	CQ22	NQ13	CQ14	NQ15	750.0 0 0.00
DIHEDral	CQ12	NQ13	CQ22	CQ16	750.0 0 180.00
DIHEDral	CQ12	NQ13	CQ22	NQ21	750.0 0 0.00
DIHEDral	CQ14	NQ13	CQ22	CQ16	750.0 0 0.00
DIHEDral	CQ14	NQ13	CQ22	NQ21	750.0 0 180.00
DIHEDral	NQ13	CQ14	NQ15	CQ16	750.0 0 0.00
DIHEDral	CQ14	NQ15	CQ16	CQ17	750.0 0 180.00
DIHEDral	CQ14	NQ15	CQ16	CQ22	750.0 0 0.00
DIHEDral	NQ15	CQ16	CQ17	NQ18	750.0 0 0.00
DIHEDral	NQ15	CQ16	CQ17	NQ19	750.0 0 180.00
DIHEDral	CQ22	CQ16	CQ17	NQ18	750.0 0 180.00
DIHEDral	CQ22	CQ16	CQ17	NQ19	750.0 0 0.00
DIHEDral	NQ15	CQ16	CQ22	NQ13	750.0 0 0.00
DIHEDral	NQ15	CQ16	CQ22	NQ21	750.0 0 180.00
DIHEDral	CQ17	CQ16	CQ22	NQ13	750.0 0 180.00
DIHEDral	CQ17	CQ16	CQ22	NQ21	750.0 0 0.00
DIHEDral	NQ19	CQ20	NQ21	CQ22	750.0 0 0.00
DIHEDral	CQ20	NQ21	CQ22	NQ13	750.0 0 180.00
DIHEDral	CQ20	NQ21	CQ22	CQ16	750.0 0 0.00
DIHEDral	PQ1	OQ23	PQ24	OQ25	750.0 0 90.00
DIHEDral	OQ23	PQ24	OQ27	CQ28	750.0 0 60.00
DIHEDral	OQ25	PQ24	OQ27	CQ28	50.0 3 180.00
DIHEDral	OQ27	CQ28	CQ29	OQ30	750.0 0 -60.00
DIHEDral	OQ30	CQ29	CQ31	OQ32	750.0 0 -90.00
DIHEDral	OQ32	CQ31	CQ33	CQ35	750.0 0 90.00
DIHEDral	CQ35	NQ36	CQ37	CQ38	750.0 0 180.00
DIHEDral	NQ36	CQ44	CQ43	CQ42	750.0 0 0.00
DIHEDral	NQ36	CQ37	CQ38	CQ39	750.0 0 180.00
DIHEDral	NQ36	CQ37	CQ38	CQ42	750.0 0 0.00
DIHEDral	CQ43	CQ42	CQ38	CQ37	750.0 0 0.00
DIHEDral	CQ37	CQ38	CQ39	NQ41	750.0 0 0.00
DIHEDral	CQ37	CQ38	CQ39	NQ40	750.0 0 180.00
DIHEDral	CQ38	CQ42	CQ43	CQ44	750.0 0 0.00

Appendices

DIHEdral	CQ42	CQ38	CQ39	NQ41	750.0	0	180.00
DIHEdral	CQ42	CQ38	CQ39	OQ40	750.0	0	0.00
DIHEdral	CQ37	NQ36	CQ44	CQ43	750.0	0	0.00
DIHEdral	CQ39	CQ38	CQ42	CQ43	750.0	0	180.00
DIHEdral	CQ35	NQ36	CQ44	CQ43	750.0	0	180.00
IMPRoper	PQ1	OQ2	OQ3	OQ4	750.0	0	35.000
IMPRoper	CQ6	CQ5	OQ7	CQ8	750.0	0	-35.000
IMPRoper	CQ8	CQ6	OQ9	CQ10	750.0	0	-35.000
IMPRoper	CQ10	CQ8	OQ11	CQ12	750.0	0	-35.000
IMPRoper	CQ12	OQ7	CQ10	NQ13	750.0	0	35.000
IMPRoper	NQ13	CQ12	CQ14	CQ22	750.0	0	0.000
IMPRoper	CQ16	NQ15	CQ17	CQ22	750.0	0	0.000
IMPRoper	CQ17	CQ16	NQ18	NQ19	750.0	0	0.000
IMPRoper	CQ22	NQ13	CQ16	NQ21	750.0	0	0.000
IMPRoper	PQ24	OQ23	OQ25	OQ26	750.0	0	35.000
IMPRoper	CQ29	CQ28	OQ30	CQ31	750.0	0	-35.000
IMPRoper	CQ31	CQ29	OQ32	CQ33	750.0	0	-35.000
IMPRoper	CQ33	CQ31	OQ34	CQ35	750.0	0	-35.000
IMPRoper	CQ35	OQ30	CQ33	NQ36	750.0	0	35.000
IMPRoper	NQ36	CQ35	CQ37	CQ44	750.0	0	0.000
IMPRoper	CQ38	CQ37	CQ39	CQ42	750.0	0	0.000
IMPRoper	CQ39	CQ38	OQ40	NQ41	750.0	0	0.000
NONBonded	PQ1	0.5849	3.3854		0.5849	3.3854	
NONBonded	OQ2	0.1591	2.8509		0.1591	2.8509	
NONBonded	OQ3	0.1591	2.8509		0.1591	2.8509	
NONBonded	OQ4	0.1591	2.8509		0.1591	2.8509	
NONBonded	CQ5	0.1200	3.7418		0.1000	3.3854	
NONBonded	CQ6	0.1200	3.7418		0.1000	3.3854	
NONBonded	OQ7	0.1591	2.8509		0.1591	2.8509	
NONBonded	CQ8	0.1200	3.7418		0.1000	3.3854	
NONBonded	OQ9	0.1591	2.8509		0.1591	2.8509	
NONBonded	CQ10	0.1200	3.7418		0.1000	3.3854	
NONBonded	OQ11	0.1591	2.8509		0.1591	2.8509	
NONBonded	CQ12	0.1200	3.7418		0.1000	3.3854	
NONBonded	NQ13	0.2384	2.8509		0.2384	2.8509	
NONBonded	CQ14	0.1200	3.7418		0.1000	3.3854	
NONBonded	NQ15	0.2384	2.8509		0.2384	2.8509	
NONBonded	CQ16	0.1200	3.7418		0.1000	3.3854	
NONBonded	CQ17	0.1200	3.7418		0.1000	3.3854	
NONBonded	NQ18	0.2384	2.8509		0.2384	2.8509	
NONBonded	NQ19	0.2384	2.8509		0.2384	2.8509	
NONBonded	CQ20	0.1200	3.7418		0.1000	3.3854	
NONBonded	NQ21	0.2384	2.8509		0.2384	2.8509	
NONBonded	CQ22	0.1200	3.7418		0.1000	3.3854	
NONBonded	OQ23	0.1591	2.8509		0.1591	2.8509	
NONBonded	PQ24	0.5849	3.3854		0.5849	3.3854	
NONBonded	OQ25	0.1591	2.8509		0.1591	2.8509	
NONBonded	OQ26	0.1591	2.8509		0.1591	2.8509	
NONBonded	OQ27	0.1591	2.8509		0.1591	2.8509	
NONBonded	CQ28	0.1200	3.7418		0.1000	3.3854	
NONBonded	CQ29	0.1200	3.7418		0.1000	3.3854	
NONBonded	OQ30	0.1591	2.8509		0.1591	2.8509	
NONBonded	CQ31	0.1200	3.7418		0.1000	3.3854	
NONBonded	OQ32	0.1591	2.8509		0.1591	2.8509	
NONBonded	CQ33	0.1200	3.7418		0.1000	3.3854	
NONBonded	OQ34	0.1591	2.8509		0.1591	2.8509	
NONBonded	CQ35	0.1200	3.7418		0.1000	3.3854	
NONBonded	NQ36	0.2384	2.8509		0.2384	2.8509	
NONBonded	CQ37	0.1200	3.7418		0.1000	3.3854	
NONBonded	CQ38	0.1200	3.7418		0.1000	3.3854	
NONBonded	CQ39	0.1200	3.7418		0.1000	3.3854	
NONBonded	OQ40	0.1591	2.8509		0.1591	2.8509	
NONBonded	NQ41	0.2384	2.8509		0.2384	2.8509	
NONBonded	CQ42	0.1200	3.7418		0.1000	3.3854	
NONBonded	CQ43	0.1200	3.7418		0.1000	3.3854	
NONBonded	CQ44	0.1200	3.7418		0.1000	3.3854	

Appendices

5.5.2 CNS Topology File for NAD⁺

```
MASS PQ1      30.97400
MASS OQ2      17.00700
MASS QQ3      17.00700
MASS OQ4      15.99900
MASS CQ5      14.02700
MASS CQ6      13.01900
MASS OQ7      15.99900
MASS CQ8      13.01900
MASS OQ9      17.00700
MASS CQ10     13.01900
MASS OQ11     17.00700
MASS CQ12     13.01900
MASS NQ13     14.00700
MASS CQ14     13.01900
MASS NQ15     15.01500
MASS CQ16     12.01100
MASS CQ17     12.01100
MASS NQ18     14.00700
MASS NQ19     15.01500
MASS CQ20     13.01900
MASS NQ21     15.01500
MASS CQ22     12.01100
MASS OQ23     15.99900
MASS PQ24     30.97400
MASS OQ25     17.00700
MASS OQ26     17.00700
MASS OQ27     15.99900
MASS CQ28     13.01900
MASS CQ29     13.01900
MASS OQ30     15.99900
MASS CQ31     13.01900
MASS OQ32     17.00700
MASS CQ33     13.01900
MASS OQ34     17.00700
MASS CQ35     13.01900
MASS NQ36     14.00700
MASS CQ37     13.01900
MASS CQ38     12.01100
MASS CQ39     12.01100
MASS OQ40     15.99900
MASS NQ41     14.00700
MASS CQ42     14.02700
MASS CQ43     13.01900
MASS CQ44     13.01900

ATOM AP       TYPE PQ1      CHARge  0.0  END
ATOM AO1      TYPE OQ2      CHARge  0.0  END
ATOM AO2      TYPE OQ3      CHARge  0.0  END
ATOM A05*     TYPE OQ4      CHARge  0.0  END
ATOM AC5*     TYPE CQ5      CHARge  0.0  END
ATOM AC4*     TYPE CQ6      CHARge  0.0  END
ATOM AO4*     TYPE OQ7      CHARge  0.0  END
ATOM AC3*     TYPE CQ8      CHARge  0.0  END
ATOM AO3*     TYPE OQ9      CHARge  0.0  END
ATOM AC2*     TYPE CQ10     CHARge  0.0  END
ATOM AO2*     TYPE OQ11     CHARge  0.0  END
ATOM AC1*     TYPE CQ12     CHARge  0.0  END
ATOM AN9      TYPE NQ13     CHARge  0.0  END
ATOM AC8      TYPE CQ14     CHARge  0.0  END
ATOM AN7      TYPE NQ15     CHARge  0.0  END
ATOM AC5      TYPE CQ16     CHARge  0.0  END
ATOM AC6      TYPE CQ17     CHARge  0.0  END
ATOM AN6      TYPE NQ18     CHARge  0.0  END
ATOM AN1      TYPE NQ19     CHARge  0.0  END
ATOM AC2      TYPE CQ20     CHARge  0.0  END
```

Appendices

```

ATOM AN3      TYPE NQ21   CHARge  0.0  END
ATOM AC4      TYPE CQ22   CHARge  0.0  END
ATOM O3       TYPE OQ23   CHARge  0.0  END
ATOM NP       TYPE PQ24   CHARge  0.0  END
ATOM NO1      TYPE OQ25   CHARge  0.0  END
ATOM NO2      TYPE OQ26   CHARge  0.0  END
ATOM NO5*     TYPE OQ27   CHARge  0.0  END
ATOM NC5*     TYPE CQ28   CHARge  0.0  END
ATOM NC4*     TYPE CQ29   CHARge  0.0  END
ATOM NO4*     TYPE OQ30   CHARge  0.0  END
ATOM NC3*     TYPE CQ31   CHARge  0.0  END
ATOM NO3*     TYPE OQ32   CHARge  0.0  END
ATOM NC2*     TYPE CQ33   CHARge  0.0  END
ATOM NO2*     TYPE OQ34   CHARge  0.0  END
ATOM NC1*     TYPE CQ35   CHARge  0.0  END
ATOM NN1      TYPE NQ36   CHARge  0.0  END
ATOM NC2      TYPE CQ37   CHARge  0.0  END
ATOM NC3      TYPE CQ38   CHARge  0.0  END
ATOM NC7      TYPE CQ39   CHARge  0.0  END
ATOM NO7      TYPE OQ40   CHARge  0.0  END
ATOM NN7      TYPE NQ41   CHARge  0.0  END
ATOM NC4      TYPE CQ42   CHARge  0.0  END
ATOM NC5      TYPE CQ43   CHARge  0.0  END
ATOM NC6      TYPE CQ44   CHARge  0.0  END

BOND AP      AO1        BOND AP      AO2        BOND AP      AO5*        BOND AP      O3
BOND AO5*    AC5*       BOND AC5*    AC4*       BOND AC4*    AO4*        BOND AC4*    AC3*
BOND AO4*    AC1*       BOND AC3*    AO3*       BOND AC3*    AC2*        BOND AC2*    AO2*
BOND AC2*    AC1*       BOND AC1*    AN9         BOND AN9     AC8          BOND AN9     AC4
BOND AC8     AN7         BOND AN7     AC5         BOND AC5     AC6          BOND AC5     AC4
BOND AC6     AN6         BOND AC6     AN1         BOND AN1     AC2          BOND AC2     AN3
BOND AN3     AC4         BOND O3      NP          BOND NP      NO1          BOND NP      NO2
BOND NP      NO5*       BOND NO5*    NC5*       BOND NC5*    NC4*        BOND NC4*    NO4*
BOND NC4*    NC3*       BOND NO4*    NC1*       BOND NC3*    NO3*        BOND NC3*    NC2*
BOND NC2*    NO2*       BOND NC2*    NC1*       BOND NC1*    NN1          BOND NN1     NC2
BOND NN1     NC6         BOND NC2     NC3         BOND NC3     NC7          BOND NC3     NC4
BOND NC7     NO7         BOND NC7     NN7         BOND NC4     NC5          BOND NC5     NC6

DIHEdral AO5*    AC5*    AC4*    AO4*    DIHEdral AC1*    AN9    AC8    AN7
DIHEdral AC1*    AN9    AC8    AN7
DIHEdral AC4     AN9    AC8    AN7
DIHEdral AC1*    AN9    AC4    AC5
DIHEdral AC1*    AN9    AC4    AN3
DIHEdral AC8     AN9    AC4    AC5
DIHEdral AC8     AN9    AC4    AN3
DIHEdral AN9     AC8    AN7    AC5
DIHEdral AC8     AN7    AC5    AC6
DIHEdral AC8     AN7    AC5    AC4
DIHEdral AN7     AC5    AC6    AN6
DIHEdral AN7     AC5    AC6    AN1
DIHEdral AC4     AC5    AC6    AN6
DIHEdral AC4     AC5    AC6    AN1
DIHEdral AN7     AC5    AC4    AN9
DIHEdral AC5     AC6    AN1    AC2
DIHEdral AN6     AC6    AN1    AC2
DIHEdral AC6     AN1    AC2    AN3
DIHEdral AN1     AC2    AN3    AC4
DIHEdral AC2     AN3    AC4    AN9
DIHEdral AC2     AN3    AC4    AC5
DIHEdral NO1     NP     NO5*   NC5*
DIHEdral NC2     NN1    NC6    NC5
DIHEdral NN1     NC2    NC3    NC7
DIHEdral NN1     NC2    NC3    NC4
DIHEDRAL NC3    NC4    NC5    NC6
DIHEDRAL NC7    NC3    NC4    NC5

```

Appendices

IMPRoper AP AO1 AO2 AO5*
IMPRoper AC4* AC5* AO4* AC3*
IMPRoper AC3* AC4* AO3* AC2*
IMPRoper AC2* AC3* AO2* AC1*
IMPRoper AC1* AO4* AC2* AN9
IMPRoper AN9 AC1* AC8 AC4
IMPRoper AC5 AN7 AC6 AC4
IMPRoper AC6 AC5 AN6 AN1
IMPRoper AC4 AN9 AC5 AN3
IMPRoper NP O3 NO1 NO2
IMPRoper NC4* NC5* NO4* NC3*
IMPRoper NC3* NC4* NO3* NC2*
IMPRoper NC2* NC3* NO2* NC1*
IMPRoper NC1* NO4* NC2* NN1
IMPRoper NN1 NC1* NC2 NC6
IMPRoper NC3 NC2 NC7 NC4
IMPRoper NC7 NC3 NO7 NN7

ACCEptor AO1 AP
ACCEptor AO2 AP
ACCEptor AO5* AP
ACCEptor AO4* AC4*
ACCEptor AO3* AC3*
ACCEptor AO2* AC2*
ACCEptor O3 AP
ACCEptor NO1 NP
ACCEptor NO2 NP
ACCEptor NO5* NP
ACCEptor NO4* NC4*
ACCEptor NO3* NC3*
ACCEptor NO2* NC2*
ACCEptor NO7 NC7

Appendices

5.6 CNS Files for Cysteine-Sulfinic Acid

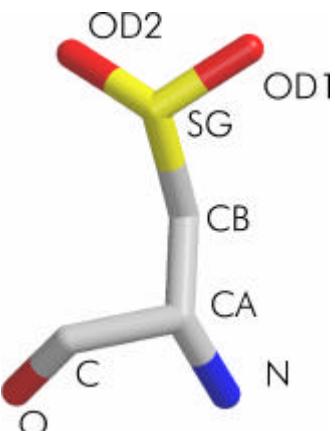


Figure 5.7 Schematic figure of cysteine-sulfinic acid (residue CSW)

5.6.1 CNS Parameter File for Cysteine-Sulfinic Acid

```
BOND NX1 CX2      1000.0  1.455
BOND CX2 CX3      1000.0  1.535
BOND CX2 CX5      1000.0  1.550
BOND CX3 SX4      1000.0  1.837
BOND SX4 OX7      1000.0  1.471
BOND SX4 OX8      1000.0  1.467
BOND CX5 OX6      1000.0  1.238

ANGLE NX1 CX2 CX3      500.0   110.41
ANGLE NX1 CX2 CX5      500.0   101.76
ANGLE CX3 CX2 CX5      500.0   111.57
ANGLE CX2 CX3 SX4      500.0   116.12
ANGLE CX3 SX4 OX7      500.0   107.89
ANGLE CX3 SX4 OX8      500.0   108.86
ANGLE OX7 SX4 OX8      500.0   113.67
ANGLE CX2 CX5 OX6      500.0   120.77

DIHEDRAL CX5 CX2 CX3 SX4      750.0  0    180.00
DIHEDRAL NX1 CX2 CX5 OX6      750.0  0    0.00
DIHEDRAL CX2 CX3 SX4 OX7      750.0  0   -60.00

IMPRPER CX2 NX1 CX3 CX5      750.0  0   -35.000
IMPRPER SX4 CX3 OX7 OX8      750.0  0   -35.000

NONBONDED NX1  0.2384  2.8509  0.2384  2.8509
NONBONDED CX2  0.1200  3.7418  0.1000  3.3854
NONBONDED CX3  0.1200  3.7418  0.1000  3.3854
NONBONDED SX4  0.0430  3.3676  0.0430  3.3676
NONBONDED CX5  0.1200  3.7418  0.1000  3.3854
NONBONDED OX6  0.1591  2.8509  0.1591  2.8509
NONBONDED OX7  0.1591  2.8509  0.1591  2.8509
NONBONDED OX8  0.1591  2.8509  0.1591  2.8509
```

Appendices

5.6.2 CNS Topology File for Cysteine-Sulfinic Acid

```
MASS NX1      15.01500
MASS CX2      13.01900
MASS CX3      14.02700
MASS SX4      32.06600
MASS CX5      13.01900
MASS OX6      15.99900
MASS OX7      17.00700
MASS OX8      17.00700

ATOM  N      TYPE NX1    CHARge  0.0  END
ATOM  CA     TYPE CX2   CHARge  0.0  END
ATOM  CB     TYPE CX3   CHARge  0.0  END
ATOM  SG     TYPE SX4   CHARge  0.0  END
ATOM  C      TYPE CX5   CHARge  0.0  END
ATOM  O      TYPE OX6   CHARge  0.0  END
ATOM  OD1    TYPE OX7   CHARge  0.0  END
ATOM  OD2    TYPE OX8   CHARge  0.0  END

BOND  N      CA          BOND  CA      CB          BOND  CA      C          BOND  CB      SG
BOND  SG     OD1        BOND  SG      OD2        BOND  C       O

DIHEDral  C      CA      CB      SG
DIHEDral  N      CA      C       O

IMPRoper  CA     N      CB      C
IMPRoper  SG     CB      OD1    OD2

ACCEptor  O      C
ACCEptor  OD1   SG
ACCEptor  OD2   SG
```

Appendices

5.7 List of GH4 enzymes

Abbreviation	Protein ID	Description	Organism
AGLA_THEMEA	O33830	α -glucosidase	<i>Thermotoga maritima</i>
AGAL_BACSU	O34645	α -galactosidase	<i>Bacillus subtilis</i>
MALH_FUSMR	O06901	Maltose-6'-phosphate glucosidase	<i>Fusobacterium mortiferum</i>
AGAL_ECOLI	P06720	α -galactosidase	<i>Escherichia coli</i>
CELF_BACSU	P46320	Probable 6-phospho- β -glucosidase	<i>Bacillus subtilis</i>
CELF_ECOLI	P17411	6-phospho- β -glucosidase	<i>Escherichia coli</i>
LPLD_BACSU	P39130	LPLD protein	<i>Bacillus subtilis</i>
GLVG_BACSU	P54716	Maltose-6'-phosphate glucosidase	<i>Bacillus subtili.</i>
ORF_1Bacillus	15613475	6-phospho- β -glucosidase	<i>Bacillus halodurans</i>
ORF_2Bacillus	15612746	6-phospho- β -glucosidase	<i>Bacillus halodurans</i>
ORF_Citrobacter	12249127	α -galactosidase	<i>Citrobacter freundii</i>
ORF_1Clostridium	15023397	Maltose-6'-phosphate glucosidase	<i>Clostridium acetobutylicum</i>
ORF_2Clostridium	15026514	6-phospho- α -glucosidase	<i>Clostridium acetobutylicum</i>
ORF_3Clostridium	18143856	Maltose-6'-phosphate glucosidase	<i>Clostridium perfringens</i>
ORF_Erwinia	13517313	Putative sugar hydrolase	<i>Erwinia rhamontici</i>
ORF_1Escherichia	12515750	Phospho- β -glucosidase; cryptic	<i>Escherichia coli O157:H7 EDL933</i>
ORF_2Escherichia	12519092	α -galactosidase	<i>Escherichia coli O157:H7 EDL933</i>
ORF_3Escherichia	12518523	Putative 6-phospho- β -glucosidase	<i>Escherichia coli O157:H7 EDL933</i>
ORF_Klebsiella	12667493	6-phospho- α -glucosidase	<i>Klebsiella pneumoniae</i>
ORF_1Listeria	16799601	Similar to 6-phospho- β -glucosidase	<i>Listeria innocua</i>
ORF_2Listeria	16799615	Similar to 6-phospho- β -glucosidase	<i>Listeria innocua</i>
ORF_3Listeria	16801516	Similar to 6-phospho- β -glucosidase	<i>Listeria innocua</i>
ORF_4Listeria	16802564	Similar to 6-phospho- β -glucosidase	<i>Listeria monocytogenes EGD-e</i>
ORF_5Listeria	16802579	Similar to 6-phospho- β -glucosidase	<i>Listeria monocytogenes EGD-e</i>
ORF_1Mesorhizobium	14026141	α -galactosidase	<i>Mesorhizobium loti</i>
ORF_2Mesorhizobium	14026146	α -galactosidase	<i>Mesorhizobium loti</i>
ORF_1Salmonella	16502875	Phospho- β -glucosidase B	<i>Salmonella enterica</i>
ORF_2Salmonella	16505287	α -galactosidase	<i>Salmonella enterica</i>
ORF_3Salmonella	16419834	Phospho- β -glucosidase	<i>Salmonella typhimurium LT2</i>
ORF_1Sinorhizobium	15141450	Probable α -galactosidase	<i>Sinorhizobium meliloti</i>
ORF_2Sinorhizobium	15141176	Probable 6-phospho- β -glucosidase	<i>Sinorhizobium meliloti</i>
ORF_1Streptomyces	6855385	Putative sugar hydrolase	<i>Streptomyces coelicolor A3(2)</i>
ORF_2Streptomyces	6900960	Putative sugar hydrolase (partial)	<i>Streptomyces coelicolor A3(2)</i>
ORF_3Streptomyces	6137043	Putative α -galactosidase	<i>Streptomyces coelicolor A3(2)</i>
ORF_patent	10059896	Sequence 8 from patent US 5985622	<i>Erwinia rhamontici</i>
ORF_Vibrio	9655770	6-phospho- β -glucosidase	<i>Vibrio cholerae</i>
ORF_Yersinia	15978271	Putative glycosyl hydrolase	<i>Yersinia pestis</i>

Appendices

5.8 Curriculum vitae

Full Name: Jacinta Anne Lodge

Date of Birth: 11.08.1975

Nationality: Australian

Education:

Primary School	1980-1982 1983 1984-1987	Lutheran Primary School, Hamilton Roslyn Primary School, Geelong Grovedale Primary School, Geelong
High School	1988-1993	Belmont High School, Geelong, Australia Achieved Victorian Certificate of Education Subjects Biology, Chemistry, Physics, Maths, Music - Solo Performance, English Score: 132/164
University	1994-1997	La Trobe University, Melbourne, Australia Bachelor of Science (Biological) with Honours Honours (Masters equivalent) Score: H1 (first class)

Career-related Employment:

01.12.96 - 31.01.97	LaTrobe University Department of Biochemistry Summer Studentship Trypsin inhibition by Anthracyclines
01.02.97 – 30-10.97	La Trobe University Department of Biochemistry Honours Student (Masters equivalent) Identification and purification of proteases from <i>Nicotiana tabacum</i>
01.05.97 - 31.02.98	LaTrobe University Department of Biochemistry Laboratory Technician Care of plant glasshouse and caterpillar colony. Dissection of caterpillars
01.03.98 - 31.03.99	Hexima, Ltd Research Assistant Agrobacterium mediated transformation of cotton plants for insect resistance. Development of ELISA-based screening technique for transformants. Production and purification of rabbit antibodies.

Appendices

20.09.99 - 30.07.00	Max-Planck Forschungsstelle "Enzymologie der Proteinfaltung" Halle/Saale Studentship Mitochondrial import. <i>In vitro</i> protein production, radioactive assays
01.08.00 - current	Freie Universität Berlin Institut für Chemie - Kristallographie PhD Student Protein Crystallography: Cloning, expression and purification of proteins for crystal production. Protein crystallisation. X-ray data collection and processing using rotating anode on X-ray generators and synchrotron sources. Building and refinement of protein structures.

Seminars and Conference Participation

- 1997 Australian Society for Biochemistry and Molecular Biology, Melbourne, Australia
1998 Lorne Protein Conference, Lorne, Australia
1999 MPI-Halle, IMB-Jena combined meeting, Wittenberg, Germany
2001 German Biochemical Society Meeting, Bochum, Germany
Poster presentation
2002 Heart of Europe Crystallography Meeting, Goslar, Germany
Talk: A new glucosidase fold
2003 DGK Meeting, Berlin, Germany
Poster presentation

Appendices

5.9 Publication of this work

Talk:

“A new glucosidase fold”

2002 Heart of Europe Crystallography Meeting, Goslar, Germany. September 25 – 27, 2002.

Poster Presentations:

Lodge, J.A., Maier, T., Raasch, C., Leibl, W. and Strater, N.

“The crystal structure of *Thermotoga maritima* α -glucosidase A defines a new clan of NAD⁺-dependent glucosidases”. Elfte Jahrestagung der Deutschen Gesellschaft für Kristallographie, Berlin, Germany. March 10 – 12, 2003

Lodge, J.A., Maier, T., Raasch, C., Leibl, W. and Strater, N.

“A new clan of glucosidases: The crystal structure of *Thermotoga maritima* α -glucosidase A”. Seventh International School on the Crystallography of Biological Macromolecules, Como, Italy. May 10 - 14, 2003.

Publication:

Lodge, J.A, Maier, T., Raasch, C., Leibl, W. and Strater, N. (2003)

“Crystal structure of family 4 *Thermotoga maritima* AglA defines a new clan of NAD⁺-dependent glucosidases”. *J. Biol. Chem.* **278** (21):19151-19158.

PDB Submission:

The atomic coordinates and structure factors (code 10BB) have been deposited in the Protein Data Bank, Research Collaboratory for Structural Bioinformatics, Rutgers University, New Brunswick, NJ. (<http://www.rcsb.org/>).