HOMOGENOUS NUCLEATION RATES OF ICE IN SUPERCOOLED BINARY LIQUID MIXTURES OF WATER + NON-ELECTROLYTES: A COMBINED THEORETICAL AND EXPERIMENTAL STUDY

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List of Symbols

- **A:** Helmholtz free energy.
- A_n : Surface area of an ice-embryo with size n.
- $\mathbf{b_0}$: Mole fraction of the non-bonded water molecules in 1 mole of hypothetical defect phase of liquid water.
- **b**₁: Mole fraction of the 1 H-bonded water molecules in 1 mole of hypothetical defect phase of liquid water.
- **b₂:** Mole fraction of the 2 H-bonded water molecules in 1 mole of hypothetical defect phase of liquid water.
- **b**₃: Mole fraction of the 3 H-bonded water molecules in 1 mole of hypothetical defect phase of liquid water.
- **b**₄: Mole fraction of the 4 H-bonded water molecules in 1 mole of hypothetical defect phase of liquid water.
- $\mathbf{c_g}$: The concentration of the critical ice nucleus in a given volume.
- ceff: Effective concentration of the critical ice nucleus, which are able to grow further in a given volume.
- **Δg*:** Free energy of activation for the diffusion of water molecules (across the ice-water interface).
- ΔG_n : Free energy of the formation of an ice-embryo with size n.
- ΔG_{n*} : Free energy of the formation of a critical ice-nucleus.
- **dr:** The thickness of the surface layer on the outer shell of an ice-embryo.
- **D:** Self diffusion constant of liquid water.
- **E**_A: Activation energy of defect formation.
- **E**_H: Apparent H-bond energy per mole of liquid water.
- **E**_i: H-bond energy in 1 mol of i-bonded water molecules.
- **f:** The net flux of water molecules to the surface of a critical ice nucleus.
- $\mathbf{f}_{\mathbf{v}}$: Molecular vibrational partition function for water molecules.
- **f_t:** Molecular translational partition function for water molecules.
- $\mathbf{f_r}$: Molecular rotational partition function for water molecules.
- **f**_i: Molecular partition function for i-bonded water molecules.
- g: Also called as Eadie's patch size. The total number of molecules included in a purely ice-like patch ignoring the molecules of hypothetical defect phase. It counts for the 4-bonded molecules in the bulk of an ice-like patch and the 2- and 3- bonded molecules on the surface of a patch.
- **G:** Gibbs free energy.
- **g2:** The number 2 H-bonded molecules on the surface of an ice-like patch before the interaction of surface molecules with the adhering non-bonded molecules of the interphase.
- **g3:** The number 3 H-bonded molecules on the surface of an ice-like patch before the interaction of surface molecules with the adhering non-bonded molecules of the interphase.
- **g4:** The number of 4 H-bonded molecules in an ice-like patch.
- $\mathbf{g}_{2\rightarrow3}$: Number of 2-bonded molecules becoming 3-bonded with the attachment of a non-bonded molecule from the interphase in between the patches.
- $\mathbf{g}_{2\rightarrow 4}$: Number of 2-bonded molecules becoming 4- bonded with the attachment of a non-bonded molecule from the interphase in between the patches.
- $\mathbf{g}_{3\rightarrow4}$: Number of 3-bonded molecules becoming 4- bonded with the attachment of a non-bonded molecule from the interphase in between the patches.
- **g(y):** Combinatorial factor in the partition function, which counts for all possible configurations for a given state of y having composed of ceratin mole fractions of different H-bonded water molecules.
- **G₂:** The number 2 H-bonded molecules on the surface of an ice-like patch after the interaction of surface molecules with the adhering non-bonded molecules of the interphase.
- **G₃:** The number 3 H-bonded molecules on the surface of an ice-like patch after the interaction of surface molecules with the adhering non-bonded molecules of the interphase.
- **G₄:** The number 4 H-bonded molecules on the surface of an ice-like patch after the interaction of surface molecules with the adhering non-bonded molecules of the interphase.
- \mathbf{h}_0 : Fractions of non-bonded water molecules in a 1 mole of liquid water.
- **h₁:** Fractions of 1-bonded water molecules in a 1 mole of liquid water.
- h_2 : Fractions of 2-bonded water molecules in a 1 mole of liquid water.
- **h₃:** Fractions of 3-bonded water molecules in a 1 mole of liquid water.
- **h₄:** Fractions of 4-bonded water molecules in a 1 mole of liquid water.
- I_z : Moment of inertia about the z axis.
- **j:** Actual number of H-bonds a water molecule forms with its neighbors, which depends on its coordination environments. It varies between 0 and 4.

J: Temperature dependent nucleation rate expressed as number of germs formed per unit volume per unit time.

J_L: Nucleation rate calculated from the slope of a linear segment which occurs left to a knick (short-time region) in an $\ln N_u/N_0$ vs V_d · t diagram.

 J_R : Nucleation rate calculated from the slope of a linear segment which occurs right to a knick (long-time region) in an ln N_u/N_0 vs V_d · t diagram.

k: Boltzmann constant.

k: Rate constant.

 n_d : The number of water molecules in a patch (or on the surface of agglomerates) which belong to the hypothetical phase of defects.

v_d: Mole fraction of hypothetical defect phase in liquid water.

n₄: Number of 4-bonded molecules in an ice-embryo.

 $n_{interface}$: Number of water molecules found in the interphase adherent on the surface of a patch.

n_s: Number of water molecules on a unit area on a patch surface.

n_T: Total number of molecules including the four-bonded molecules in the iner-shell, and defect molecules on the surface, in an ice-embryo.

N: Total number of water molecules in a system.

N_{A:} Avagadro's number.

N_{Agg.-S}: Number of 4-bonded agglomerates (or ice embryos) placed at the surface of a patch.

 N_{Agg-T} : Total number of four-bonded agglomerates present in a patch.

 N_c : Number of water molecules in contact with the unit surface area of an ice embryo.

 N_i : The number of i-bonded water molecules in 1 mole of liquid water.

m: Molecular mass of a water molecule.

μ_{liq}: Chemical potential, i.e. Molar Gibbs free energy of liquid water.

 μ_{ice} : Chemical potential, i.e. Molar Gibbs free energy of ice.

 μ_d : Chemical potential, i.e. Molar Gibbs free energy of the hypothetical defect phase.

 $\Delta \mu_{ice}$ -liq (or $\Delta \mu$): μ_{ice} - μ_{liq} **5**: Symmetry number.

 $\sigma_{i/w}$: Surface tension of the ice-water interface on a growing ice embryo.

P_{diff}: The probability that a molecule will diffuse out is lattice point.

p: H-bond formation probability in the hypothetical phase of lattice defects in liquid water.

p: Pressure.

 P_{rel} (n): Probability of finding an ice embryo of size n in a patch

 $\rho_{T}(liq)$: Density of liquid water at a given temperature T.

 ρ_{T} (ice): Density of ice at a given temperature T.

 $\rho_T(\mathbf{d})$: Density of hypothetical defect phase at a given temperature T.

 $\mathbf{r}_{inner-shell}$: Radius of the inner shell of a patch below its adherent interphase.

 \mathbf{r}_{patch} : Radius of a patch having a spherical geometry.

 $\hat{\mathbf{R}}_{agg}$: The radius of the inner-shell of an ice-embryo.

r(T): Van der Waals radius of the water molecule in ice at a given temperature.

s: Total number of assigned vibrational frequencies for a water molecule, which depends on its H-bonding state.

S: Entropy.

S_T: Total patch size, i.e. Total number of 4-bonded molecules and the molecules with less than 4-Hydrogen bonds which constitutes a patch and its adherent interface.

 t_{M} : Lifetime of the metastable state in liquid water.

 t_R : The relaxation time of the metastable state in liquid water.

<τ>: Average mean passage time for the water molecules to leave a surface layer of certain thickness.

T: Temperature

U: Internal energy.

 χ_{H} : The electronegativity of the hydrogen atom.

X_i: Mole fraction of ice-like phase in liquid water.

 χ_0 : The electronegativity of the oxygen atom.

V_d: Packing volume of a water molecule that belong to the hypothetical defect phase.

 V_f : Free volume of a water molecule in a translational motion.

 $V_m(d)$: Molar volume of the hypothetical defect phase.

 V_{ice} : Packing volume of a water in ice.

 V_m (ice): Molar volume of ice.

 Ω_{eff} :

 V_{patch} : Total volume of a patch assuming a spherical geometry.

 V_{surf} : The volume of the surface layer of with a thickness dr for a spherical agglomerate.

 \mathbf{W}_{eff} : Total effective surface area available for the growth of four-bonded agglomerates (ice-embryos) on the surface of a patch.

The effective surface area for the growth of an agglomerate, which lies near to the surface.

 Ω_g : The surface area of the critical ice nucleus.

 $\mathbf{W}_{\text{patch}}^{\text{s}}$: Surface area of the inner shell of a patch just below its adherent interface.

z: Maximum number of H-bond that water can form with its nearest neighbors in liquid water (z = 4).

Z: Canonical partition function of liquid water.

Z: The Zeldovitch factor

Z(avg): Canonical partition function of an ice embryo, with the most probable size.

Z(n): Canonical partition function of an ice embryo, which consists of n water molecules.